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the amount of that substance required to form a saturated solution with a specific amount of solvent at a specified temperature. Hydration ...

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1- Types of Mixtures Solutions are
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are homogeneous mixtures of two or
more substances in a single phase.
Soluble describes a substance as

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capable of being dissolved. Solvent is the dissolving medium in a solution. Solute is the substance that is dissolved

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Solutions. Section 1- Types of Mixtures
Solutions are homogeneous mixtures of

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two or more substances in a single phase. Soluble describes a substance as capable of being dissolved. Solvent is the dissolving medium in a solution.

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ANSWER 1. c 2. a 3. b 2. a. alcohol b.
water c. the gels 3. The mixture is a
colloid. The properties are consistent
with those reported in Table 3 on page
404 of the text. The

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Mixtures The Solution Process
Concentrations of Solution 1 2 Section
12.1 Types of Mixtures Chapter 12
Section 1 Types of Mixtures p. 401-406 2

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3 Vocabulary Colloid Dispersed phase
Dispersing medium Electrolyte
Nonelectrolyte Solute ...

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12 REVIEW Solutions Teacher Notes and
Answers Chapter 12 SECTION 1 SHORT

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ANSWER 1. c 2. a 3. b 2. a. alcohol b. water c. the gels 3. The mixture is a colloid. The properties are consistent with those reported in Table 3 on page 404 of the text. The particle size is small, but not too small, and the mixture

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CHAPTER 12 REVIEW. Solutions.

SECTION 1. SHORT ANSWER Answer the following questions in the space provided. 1. Match the type of mixture on the left to its representative particle diameter on the right. c

solutions(a)larger than 1000 nm. a

suspensions(b)1 nm to 1000 nm. b

colloids(c)smaller than 1 nm.

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Solution Process Concentrations of
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Chemistry Chapter 12-Solutions Section
1- Types of Mixtures Solutions are
homogeneous mixtures of two or more
substances in a single phase. Soluble
describes a substance as capable of
being dissolved. Solvent is the dissolving
medium in a solution. Solute is the

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substance that is dissolved in a solution.

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